

Graphical Abstracts/J. Fluorine Chem. 128 (2007) 885–888

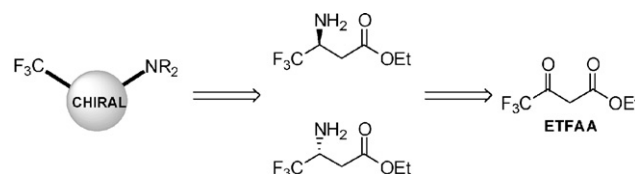
J. Fluorine Chem., 128 (2007) 889

Ethyl-4,4,4-trifluoroacetoacetate (ETFAA), a powerful building block for enantiopure chirons in trifluoromethyl- β -amino acid series

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Jean-Christophe Plaquevent^a

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New transformations of ethyl-4,4,4-trifluoroacetoacetate (ETFAA) give access to a series of enantiopure chirons bearing both a trifluoromethyl group and an amino moiety. The key intermediate is obtained optically pure by a resolution procedure.

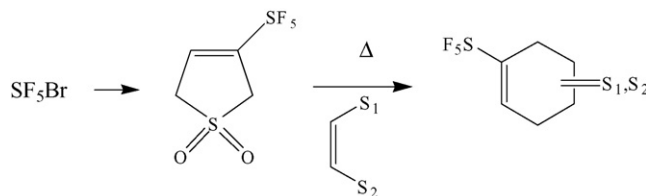
J. Fluorine Chem., 128 (2007) 896

Synthesis and reactions of 2-SF₅-butadiene

Rolf W. Winter, Gary L. Gard

Portland State University, Portland, OR 97207-0751, USA

A stable and storable precursor of 2-SF₅-butadiene, 3-SF₅-3-sulfolen, has been synthesized and its reactivity studied with several olefinic compounds.



J. Fluorine Chem., 128 (2007) 902

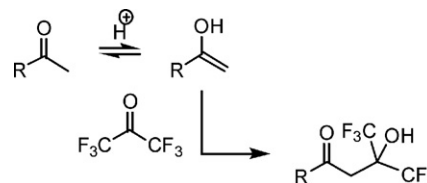
The cross-aldol reaction of hexafluoroacetone (HFA) with ketones catalyzed by an acid

Takeo Komata^{a,b}, Kei Matsunaga^b, Yoshiki Hirotsu^b, Shinya Akiba^b,
Katsuyuki Ogura^a

^aDiversity and Fractal Science, Graduate School of Science and Technology, Chiba
University, 1-33 Yayoi-cho, Inage-ku, Chiba-shi, Chiba 263-8522, Japan

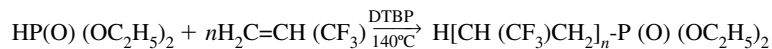
^bChemical Research Center, Central Glass Co. Ltd., 2805 Imafuku-Nakadai, Kawagoe-shi,
Saitama 350-1151, Japan

The cross-aldol reaction of hexafluoroacetone (HFA) and ketones using an acid catalyst occurred efficiently.



J. Fluorine Chem., 128 (2007) 910

Radical telomerization of 3,3,3-trifluoropropene with diethyl hydrogen phosphonate: Characterization of the first telomeric adducts and assessment of the transfer constants

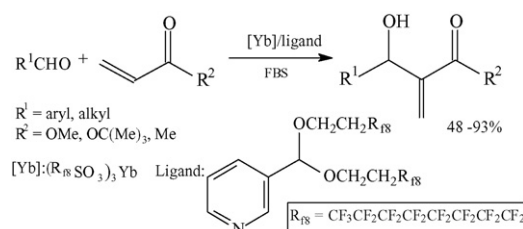
George Kostov^a, Bruno Ameduri^a, Stephan M. Brandstadter^b^aIngénierie et Architectures Macromoléculaire, Institut Charles Gerhardt Ecole Nationale Supérieure de Chimie de Montpellier, UMR CNRS 5253, 8 rue de l'École Normale, F 34296 Montpellier Cedex 5, France^bChemtura, 1801 US Highway 52 N.W., West Lafayette, IN 47996-2200, USAThe transfer constant value, C_T^n , for HP(O)(OEt)₂ is 0.75 at 140 °C.

J. Fluorine Chem., 128 (2007) 919

A novel ytterbium/perfluoroalkylated-pyridine catalyst for Baylis–Hillman reaction in a fluororous biphasic system

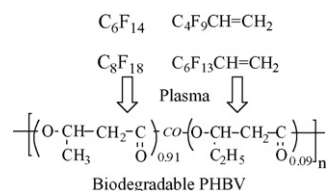
Wen-Bin Yi, Chun Cai, Xin Wang

Chemical Engineering College, Nanjing University of Science & Technology, Nanjing 210094, China

Ytterbium perfluorooctanesulfonate [Yb(OPf)₃] catalyses the highly efficient Baylis–Hillman reaction in the presence of a catalytic amount of a novel perfluoroalkylated-pyridine as a ligand in a fluororous biphasic system (FBS).

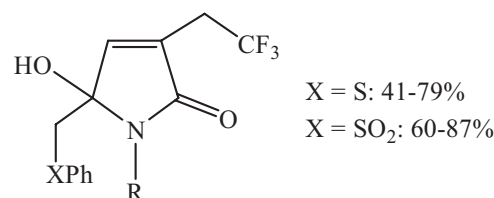
J. Fluorine Chem., 128 (2007) 925

Fluorinated-plasma coating on polyhydroxyalcanoate PHBV. Effect on the biodegradation

Nadia Guerrouani^a, Audrey Baldo^a, Taïb Maarouf^a, Anna M. Belu^b, Camille M. Kassis^b, André Mas^a^aInstitut C. Gerhardt, UMR 5253, ESOM, Université Montpellier II, Place Eugène Bataillon, 34095 Montpellier Cedex 5, France^bDepartment of Chemistry CB#3290 Venable and Kenan Laboratories, University of North Carolina, Chapel Hill, NC 27599-3290, USAA hydrophobic coating ($\gamma_s = 18.8 \text{ mJ m}^{-2}$) from plasma-polymerization of saturated and unsaturated fluorinated monomers, that delays the biodegradation of PHBV.

J. Fluorine Chem., 128 (2007) 931

Synthesis of new 3-(2,2,2-trifluoroethyl)-5-hydroxy-5-(phenylsulfanyl- or phenylsulfonyl-methyl)-1,5-dihydropyrrol-2-ones starting from α,β -unsaturated γ -lactones and primary amines

Jean-Philippe Bouillon^a, Yuriy Shermolovich^{a,b}, Sergiy Mykhaylychenko^{a,b}, Dominique Harakat^c, Bernard Tinant^d^aLaboratoire Sciences et Méthodes Séparatives (SMS), EA 3233, Université de Rouen, IRCOF, F-76821 Mont-Saint-Aignan Cedex, France^bInstitute of Organic Chemistry, National Academy of Sciences of Ukraine, 5, Murmanska, 02094 Kiev, Ukraine^cLaboratoire Réactions Sélectives et Applications, UMR CNRS 6519, Université de Reims, Faculté des Sciences, B.P. 1039, F-51687 Reims Cedex 2, France^dUnité de Chimie Structurale et des Mécanismes Moléculaires, Université Catholique de Louvain, 1 Place Louis Pasteur, B-1348 Louvain-la-Neuve, Belgium

X = S: 41-79%

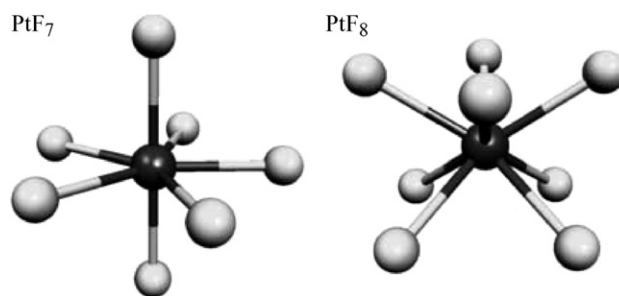
X = SO₂: 60-87%

J. Fluorine Chem., 128 (2007) 938Platinum fluorides beyond PtF₆?

Sebastian Riedel

Department of Chemistry, University of Helsinki, A.I. Virtasen aukio 1, FIN-00014 Helsinki, Finland

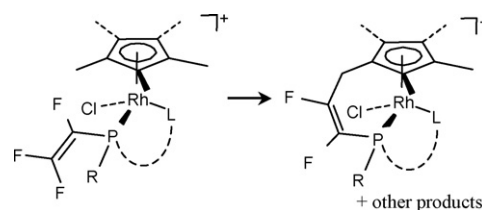
While the highest platinum fluoride characterized beyond doubt is PtF₆, we have investigated higher platinum fluorides by state-of-the-art quantum-chemical calculations up to high oxidation state +VIII.

*J. Fluorine Chem.*, 128 (2007) 943

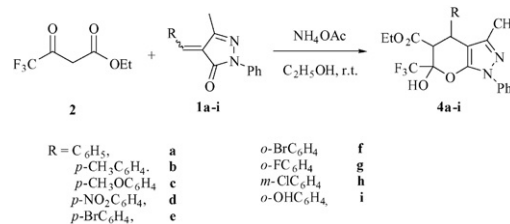
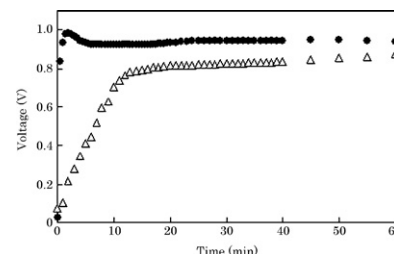
Pentamethylcyclopentadienyl rhodium(III) trifluorovinyl phosphine complexes and attempted intramolecular dehydrofluorinative coupling of pentamethylcyclopentadienyl and trifluorovinyl phosphine ligands

Nicholas A. Barnes^a, Alan K. Brisdon^a, Mark Nieuwenhuyzen^b, Robin G. Pritchard^a, Graham C. Saunders^b^aSchool of Chemistry, University of Manchester, Manchester M13 9PL, United Kingdom^bSchool of Chemistry and Chemical Engineering, Queen's University Belfast, David Keir Building, Belfast BT9 5AG, United Kingdom

Treatment of cationic rhodium piano stool complexes with trifluorovinylphosphines with proton sponge yields complexes of tethered ligands.

*J. Fluorine Chem.*, 128 (2007) 952Regioselective synthesis of 6-trifluoromethyl-1,4,5,6-tetrahydropyrazolo[3,4-*b*]pyran derivativesDongmei Li^a, Liping Song^{a,b}, Shaodi Song^a, Shizheng Zhu^b^aDepartment of Chemistry, School of Science, Shanghai University, No. 99, Shangda Road, Shanghai 200444, China^bKey Laboratory of Organofluorine Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 354 Fenglin Lu, Shanghai 200032, China

A facile two-step procedure for synthesis of 6-(trifluoromethyl)-1,4,5,6-tetrahydropyrazolo[3,4-*b*]pyran (**4**) derivatives from 1-phenyl-4-arylidene-5-pyrazolones (**1**) with ethyl trifluoroacetoacetate (**2**) is presented. To increase the efficiency of this reaction the one-pot process was also developed, with little lower yield. Treatment of **4** with P₂O₅ or conc. H₂SO₄ could not give the dehydrated products.

*J. Fluorine Chem.*, 128 (2007) 958Preparation of COF₂ using CO₂ and F₂ in the electrochemical cell with PbSnF₄ as a solid electrolyteYasuo Hasegawa^b, Akitomo Nagasaka, Kim Jae-Ho, Susumu Yonezawa, Masayuki Takashima^a^aDepartment of Materials Science and Engineering, Faculty of Engineering, Fukui University, 3-9-1 Bunkyo, Fukui 910-8507, Japan^bForensic Science Laboratory, Fukui Prefectural Police Headquarters, 3-17-1 Ote, Fukui 910-8515, JapanEmf with anode gas of 0.1 MPa CO₂ and cathode gas of F₂ and argon.

Synthesis and polymerization of novel fluorinated acrylates and methacrylates bearing alkoxy groups derived from radical addition reaction of perfluoroisopropenyl ester

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^aDepartment of Applied Chemistry, Graduate School of Engineering, Saitama Institute of Technology, 1690 Fusaiji, Fukaya 369-0293, Japan

^bDepartment of Life Science and Green Chemistry, Saitama Institute of Technology, 1690 Fusaiji, Fukaya 369-0293, Japan

Radical addition of 2-benzoxypentafluoropropene [$\text{CF}_2=\text{C}(\text{CF}_3)\text{OCOC}_6\text{H}_5$] (BPFP) with alcohols such as ethanol and 2-propanol was investigated to afford fluorinated alcohols. Radical addition of BPFP with cyclic ethers such as tetrahydrofuran, 1,3-dioxolane and tetrahydropyran was also achieved to afford addition products followed by hydrolysis to yield fluorinated alcohols possessing cyclic structures. Novel fluoroalkyl acrylates and methacrylates were synthesized from the fluorinated alcohols with (meth)acryloyl chlorides. Radical polymerization of the fluoroalkyl (meth)acrylates yielded polymers of 1.2×10^5 as the highest molecular weight.

